Title: Honeaite, a new gold-thallium-telluride from the Eastern Goldfields, Yilgarn Craton, Western Australia.

Running title: Honeaite, a new gold-thallium-telluride

Abstract

Introduction

Regional geological setting of the Karonie gold deposits

General geology of the Karonie gold deposits

Gold mineralisation

Honeaite

Occurrence

Appearance and physical properties

Chemistry

Crystallography

Discussion

Paragenetic position of honeaite

Comparisons with other telluride-bearing deposite in the Eastern Goldfields and classification of the mineralisation

Acknowledgements

References

Corresponding author: Clive M Rice, Department of Geology & Petroleum Geology, School of Geosciences, Meston Building, Kings College, University of Aberdeen, Aberdeen AB24 3UE, United Kingdom.

Email, c.rice@abdn.ac.uk

Phone, 44 1467 651218

Fax, no fax
Honeaite, a new gold-thallium-telluride from the Eastern Goldfields, Yilgarn Craton, Western Australia

Clive M Rice\textsuperscript{1}, Mark D Welch\textsuperscript{2}, John W Still\textsuperscript{1}, Alan J Criddle\textsuperscript{1} & Chris J Stanley\textsuperscript{2}

\textsuperscript{1}Department of Geology & Petroleum Geology, School of Geosciences, Meston Building, Kings College, University of Aberdeen, Aberdeen AB24 3UE, United Kingdom.

\textsuperscript{2}Department of Earth Sciences, Natural History Museum, Cromwell Road, London SW7 5BD, United Kingdom.

† mortuus est 2nd May 2002.

*E-mail: c.rice@abdn.ac.uk
Honeaite, ideal formula \( \text{Au}_3\text{TI}_{1.01}\text{Te}_{2.00} \), is a new mineral from the late Archaean Karonie gold deposit, Eastern Goldfields province, Western Australia. Honeaite is found with native gold, (omission), tellurobismuthite, petzite, hessite, calaverite, melonite, mattagamite, frohbergite, altaite, (omission)pyrrhotite and molybdenite. These minerals are concentrated in microvughs and microfractures mainly within areas of prehnite alteration of amphibolite. The mineralisation appears to have been deposited under greenschist facies conditions at lower temperatures than most gold deposits in the Eastern Goldfields.

Single-crystal X-ray studies identified the structure of honeaite as orthorhombic, space group \( Pbcm \), with \( a = 8.9671 \) (4)Å, \( b = 8.8758(4) \) Å, \( c = 7.8419(5) \) Å, giving \( V = 624.14(6) \) Å\(^3\) with \( Z = 4 \). The strongest reflections of the calculated powder X-ray diffraction pattern are \([d_{\text{rel}}(hkl)]\): 2.938(100)(022), 2.905 (39,8)(322, 411), 2.989 (31)(300), 2.833 (23)(310), 1.853 (17)(332). Electron microprobe analysis (EDS mode) gave (wt%) Au 56.33, Tl 19.68, Te 24.30, total 100.31, leading to an empirical formula (based on 2 Te apfu) of \( \text{Au}_3\text{TI}_{1.01}\text{Te}_{2.00} \). Honeaite is black with a metallic lustre and no observed cleavage. The calculated density is 11.18 g/cm\(^3\). In reflected plane-polarized light it is slightly bluish grey. Between crossed polars it is weakly anisotropic with dark brown to dark blue rotation tints. Reflectance values in air and in oil are given.

Honeaite is named after the late Russell M. Honea (1929-2002). Omission.

**KEYWORDS:** Honeaite; new mineral; gold-thallium-telluride; Eastern Goldfields; Western Australia.

---

Rice et al.
Introduction

Honeaite was discovered by the late Russell M. Honea in core from the Karonie gold deposit in the Eastern Goldfields of Western Australia at \( \sim 31^\circ02'08''S \) (latitude), \( 122^\circ33'39''E \) (longitude). He sent the samples containing what he believed to be a new Au-Tl-Te-bearing mineral to the late Richard A. Kosnar, a mineral dealer and owner of the company Mineral Classics located near Black Hawk, Colorado, USA. Kosnar passed them on to CMR in 1999. The chemistry and optics were determined at an early stage in the study and supported Honea’s suggestion that the mineral was new to science. However, completion of the study was delayed until the discovery in 2014 of grains large enough for a full structural analysis.

The Karonie deposit is on Cowarna Downs Station, which is 105 km east of Kalgoorlie and 7 km south of Karonie Siding on the Trans Australian Railway (Fig. 1.). It was initially managed by a joint venture between Freeport-McMoRan Australia Ltd., Karonie Gold N.L. and Golconda Minerals and later by Poseidon Gold Ltd.. Further exploration work has been carried out since (Jones 2007). Gold was produced from a series of open pits and some underground workings, with a total combined production of 4960 kg Au during the period 1987 to 1992 (Roberts et al. 2004).

It is appropriate to name the mineral in honour of the late Russell M Honea (1929-2002) who was a well-known mineralogist in the western USA and, indeed, globally. Russ Honea worked with some of the great mineralogists of the last century such as Cliff Frondel and Connie Hurlbut Jrn.. He was initially a professor based at the University of Colorado, Boulder, Colorado, USA and later worked as an independent consulting geologist. In addition to redefining the tellurides empressite and stuetzite (Honea, 1964) and describing several new minerals (e.g. billingsleyite, Frondel and Honea, 1968; chambersite, Honea and Beck, 1962) among his many publications, he also worked on the original description of the Karonie mineralogy for the mining company Freeport-McMoRan Inc.. He donated substantial numbers of specimens from his collection to the mineralogical museums of the University of Arizona and to Denver Museum of Nature and Science.

The mineral and name were approved by the Commission on New Minerals, Nomenclature and Classification of the International Mineralogical Association (IMA 2015-060). Holotype material is deposited in the collections of the Natural History Museum, London, catalogue number BM 2015, 36.

Honeaite appears to be identical to Unnamed Mineral 1993-27 Te:AuTl (Smith and Nickel, 2007). Chemical data for this unnamed mineral are given in Nechaev & Bondarenko (1993) and Bondarenko et al. (1993). Nechaev & Cook (2000) describe \( \text{Au}_3\text{Te}_2 \) from the Maiskoe deposit, Ukraine while Bondarenko et al. (2005) describe \( \text{Au}_3\text{TI}_2\text{Te}_2 \) from the Potashnya deposit, Ukraine. The crystal structure of none of these phases was determined.

Regional geological setting of the Karonie gold deposits

The Karonie deposit occurs near the southeastern margin of the Eastern Goldfields Superterrane of the Archaean Yilgarn Craton (Fig. 1). The Superterrane is composed of a...
collage of fault-bounded, late Archaean (2.73-2.65 Ga) greenstone belts separated and
intruded by extensive tracts of granitoid batholiths. It is divided into a number of terranes,
which include the Kurnalpi terrane in which the Karonie deposit is located (Fig. 1) (Swager,
1997; Cassidy et al. 2006; Pawley et al. 2012). The Superterrane has had a complex tectonic
history (eg Swager 1997; Blewett et al. 2010) and six stages of deformation have been
distinguished by Blewett et al. (op cit). Regional metamorphic grades range from greenschist
to amphibolite facies. Peak metamorphism coincided with D2 of Blewett et al. (2010) and
was probably contemporaneous with the bulk of granite emplacement at c. 2.6Ga (Jones
and Hall, 2004 and numerous references therein).

The Yilgarn Craton is famous for its gold deposits, most of which are found on the eastern
side, the so-called Eastern Goldfields. The largest by far and a world class gold deposit is the
Kalgoorlie Golden Mile (1500 t Au produced). Most of the gold is believed to have been
emplaced during late transpressional events (D4 of Blewett et al. 2010), shortly before
cratonization (Vielreicher et al. 2015). A final relatively minor stage of telluride-rich
mineralization was controlled by strike-slip faulting (DS of Blewett et al., 2010).

General Geology of the Karonie gold deposits

Karonie is located within the Murrin domain near the eastern margin of the Kurnalpi terrane
(Fig. 1)(Cassidy et al. 2006; Jones 2007). The domain stratigraphy consists of a
metamorphosed sequence of andesite-dominated volcanic and volcaniclastic rocks and
interlayered metasediments, which were emplaced in an oceanic intra-arc setting (Barley et
al. 2008). The deposits are about 3 km from the contact with the Erayinia Granitic Suite
(Jones 2007).

The Karonie gold deposits lie within the north-trending Karonie shear zone which is over
100m wide. The gold deposits are predominantly hosted by quartz amphibolite and, in the
northern part of the deposit, minor quartz-biotite-rich metasediments. The largest deposit,
where honeaite was discovered, is the Main Zone orebody which has a strike length of about
600m and is about 40m wide (Fig. 2). It consists of several lenses that are generally
conformable with the strike of the shear zone (Pigott and Green 1990; Poseidon Exploration
Ltd., 1992; de Luca 1995; Roberts et al. 2004).

The metamorphic grade in the shear zone is mid to upper amphibolite facies whereas
either side of the shear zone the grade decreases to greenschist facies actinolite-bearing
rocks. There is intense ductile deformation in the shear zone and later brittle and brittle-
ductile faults (de Luca 1995; Roberts et al., 2004). The two most accessible and informative
accounts of the Karonie gold deposits are provided by de Luca (1995) and Roberts et al.
(2004).

Gold mineralization

Four different types of alteration have been identified in the Karonie deposit (Roberts et al.
2004). In chronological order these are biotite-rich assemblages, mafic gneiss (coarsely
banded hornblende-rich rocks), calc-silicate (thin pyroxene-rich veins with alteration
selvedges in which hornblende has been replaced by diopside, plagioclase by epidote and
ilmenite by titanite) and late alteration (biotite, calcic plagioclase and quartz) related to
brittle-ductile faults (Roberts et al. 2004). In addition to the above, lower temperature
alteration minerals including sericite, especially replacing feldspar, chlorite and prehnite are
variably developed in all assemblages (de Luca 1995; Roberts et al. 2004).

Pigott and Green (1990) report a spatial association between gold mineralisation and calc
silicate alteration, brittle-ductile faults and amphibolite/metasediment contacts. However,
Roberts et al (2004) report that the mineralization is found also in mafic gneiss and biotite-
bearing mafic gneiss and that there is no consistent association with any rock type. In Figure
2 the mineralization forms a strike-parallel lense and is hosted by four rock types. De Luca
(1995) noted the association of gold with low temperature minerals such as tellurides,
epidote and prehnite.

Previous studies have reported that gold is the dominant component in clouds of fine
grained ore minerals in hornblende and especially patches of epidote-clinozoisite and
prehnite alteration (Pigott and Green, 1990; de Luca, 1995). Coarser gold occurs along
fractures and cleavage planes in hornblende. Ore minerals associated with the gold are
various Au-Ag-Ni-Pb-Bi-bearing tellurides, native Te and molybdenite (Pigott and Green
1990; de Luca 1995). Pigott and Green (1990) were the first to report an unnamed Tl
telluride, undoubtedly honeaite and discovered by Honea, who was a consultant at the
mine. Up to 10% pyrrhotite, pyrite and chalcopyrite occurs in all rock types and is unrelated
to gold mineralization.

Honeaite

Occurrence

The core samples containing honeaite came from borehole KD 41 at 78m, where the
sequence is dominated by quartz amphibolite (Poseidon Exploration Ltd. 1992). The
borehole was collared on section 10280N in the northern part of the Main Zone ore body
(Fig. 2). Further honeaite-bearing material from the same core interval was provided by Ted
Wilton (ex Exploration Manager-Western Australia for Freeport of Australia) and was
included in this study. In the following description of the occurrence of honeaite the
compositions by electron microprobe of associated silicates and ore minerals are provided
in Table 1.

The host rock to the mineralization is a banded amphibolite with a primary assemblage
consisting predominantly of randomly orientated sheaves of ferro-hornblende and mosaic
textured calcic plagioclase (An 31) with small amounts of orientated ilmenite and pyrite and
traces of zircon (omission Table 1).

Albitisation and calc-silicate alteration are widespread in the honeaite-bearing samples
(Fig. 3A, B). Calcic plagioclase is replaced by more sodic plagioclase (An 10) along grain
boundaries and in larger patches (omission Table 1). Sodic plagioclase commonly contains finely disseminated sericite.

Calc-silicate alteration includes actinolite, epidote, prehnite and titanite. Actinolite of variable iron content occurs as patches of alteration within and around the edges of hornblende (Fig. 4; Table 1). Epidote and prehnite are widely developed as replacement products of hornblende and sodic plagioclase respectively (Fig. 4, omission Table 1).

Prehnite also replaces epidote and cements fractures in epidote and mantles actinolite (Figs. 4, 5, omission). There are small amounts of chlorite replacing hornblende and of apatite associated with epidote and prehnite. Late brittle fractures are filled by prehnite, sometimes with a little chlorite, and these may cut earlier prehnite (Fig. 6). Titanite, like prehnite, also occurs as late brittle veins and as a replacement product of ilmenite (Figs. 5, 6).

Honeaite and other ore minerals, form clouds of small (largest c. 300 microns, most < 100 microns) inclusions roughly following the metamorphic banding (Fig. 3A). The mineralization also shows a broad spatial relationship with areas of epidote and prehnite alteration as noted by de Luca (1995) in a comprehensive study (Fig. 3A). However, in detail the mineralization shows a much closer spatial relationship to areas of prehnite (Figs. 6, 7). Given the small number of samples in our study we cannot say whether this is true of the deposit as a whole.

The inclusions consist mainly of gold (low Ag) and tellurobismuthite with small amounts of molybdenite, petzite, hessite, calaverite, melonite, mattagamite, frohbergite, altaite, pyrrhotite and honeaite. There are also composite grains of these minerals and honeaite, variously with gold, petzite, tellurobismuthite, molybdenite, chalcopyrite and pyrrhotite (Figs. omission, 8, 9) (Table 1). This mineral assemblage and its style are very close to that described from Karonie by Pigott and Green (1990) and de Luca (1995). Late cross-cutting prehnite veinlets contain a few grains of gold and a bismuth telluride but are largely barren (Fig. 6). Small amounts of scheelite occur in a heavy mineral concentrate from the same core interval but it has not been observed in situ. It probably accounts for the W anomalies described in the deposit by de Luca (1995).

Over 25 honeaite grains were found occurring in microvughs and microfractures predominantly in prehnite with a few in epidote and hornblende (Figs. 9 to 13). Honeaite morphology is determined by the shape of the vugh; no undoubted crystal faces were observed. The other tellurides and associated sulphides are also mainly vugh- and fracture-controlled within prehnite.

**Appearance and physical properties**

Honeaite is black with a metallic lustre and no observed cleavage. The calculated density is 11.18 g/cm$^3$. In plane-polarized incident light, honeaite is slightly bluish grey in colour (Figure 13), very weakly bireflectant and very weakly pleochroic from grey ($R_2$) to slightly bluish grey ($R_1$). The mineral does not show any internal reflections. Between crossed polars, honeaite is weakly anisotropic with dark brown to dark blue rotation tints.
Reflectance data were obtained in air and in oil by the late Alan Criddle using the instrumentation and techniques described in Stanley et al (2002). The data are given in Table 2 and plotted in Figure 14. Readings were taken for specimen and standard (WTIC) maintained under the same focus conditions.

Chemistry

Electron microprobe data were obtained on 17 grains of honeaite (Table 3) omission. These data were subsequently checked using a wavelength dispersive microprobe and suitable standards with excellent agreement between the datasets. The empirical formula (based on 2 Te apfu) is \( \text{Au}_{3.00} \text{Tl}_{1.01} \text{Te}_{2.00} \) giving a simplified formula of \( \text{Au}_3 \text{TlTe}_2 \) which requires (wt%) Au 56.25, Tl 19.46 and Te 24.29, total 100.00.

Crystallography

Full details of the crystal structure of honeaite are reported elsewhere (Welch et al. submitted), so only a brief summary is given here. Honeaite is orthorhombic, space group \( Pbcm \), with unit-cell parameters \( a = 8.9671(4) \) Å, \( b = 8.8758(4) \) Å, \( c = 7.8419(5) \) Å, \( V = 624.14(6) \) Å\(^3\) (\( Z = 4 \)). The structure has been solved and refined using SHELX (Sheldrick, 2008) to final agreement indices \( R_1 = 0.033, wR_2 = 0.053 \), Goodness-of-Fit = 1.087, for full anisotropic refinement. The structure topology is completely novel and is composed of two components: (i) corrugated double-sheets of six-membered rings of corner-linked TeAu\(_3\) pyramids, with Te and Au atoms located at apices, and Te having the one-sided three-fold coordination that is characteristic of a stereoactive lone-pair; there is additional intra-sheet connectivity via Au-Au bonds. (ii) rows of octahedrally-coordinated Tl atoms lying in the grooves of the corrugated sheets. Connections between adjacent TeAu\(_3\) double-sheets involve only Tl-Au bonds. The structure is shown in Figure 15. A CIF file containing data collection and structural information is deposited with the journal.

The crystals were too small to collect a quasi-Gandolfi `powder` pattern from a rotating single crystal and so the reflections were calculated based on the crystal structure (Table 4).

The only compositionally similar Au-bearing phases of which we are aware are synthetic CsAu\(_3\)S\(_2\), RbAu\(_3\)Se\(_2\) and CsAu\(_3\)Se\(_2\) (Klepp and Weithaler, 1996). The structure topology shared by these compounds is simple and consists of alternating planar sheets of SeAu\(_3\) (SAu\(_3\)) pyramids and Cs (Rb) atoms, and is very different from the corrugated sheet of honeaite. Most importantly, the coordination of Se and S atoms is very different from Te in honeaite: Se and S are octahedrally-coordinated to three Au and three Cs (Rb) cations, e.g. Se[Au\(_3\)Cs\(_3\)] in CsAu\(_3\)Se\(_2\).

One of the two non-equivalent Te atoms of honeaite has possible bonding distances to Tl, whereas the other Te is only bonded to Au. In both cases, Te[Au\(_3\)Tl\(_2\)] and Te[Au\(_3\)], the coordination is highly asymmetric (one-sided), suggesting steroeactivity of both Te atoms.
For CsAu$_3$Se$_2$ and CsAu$_3$S$_2$, the Cs-Se and Cs-S distances of 3.76 and 3.73 Å, respectively, imply that the Te(1)-Tl distance at 3.49 Å in honeaite is likely to be a bond, albeit contributing little bond valence to Te. Inclusion of much longer Te-Tl distances in honeaite (> 3.9 Å) are very unlikely to be bonds as they would result in very unusual coordination for Te. Thus, Tl can be considered to be octahedrally-coordinated. This coordination environment contrasts significantly with that of Cs and Rb in CsAu$_3$S$_2$, RbAu$_3$Se$_2$ and CsAu$_3$Se$_2$, in which they are octahedrally-coordinated by six Se (S) atoms.

The minerals petzite Ag$_3$AuTe$_2$ (Frueh, 1959) and fischesserite Ag$_3$AuSe$_2$ (Bindi and Cipriani, 2004) have the same 3:1:2 stoichiometry but very different structures from that of honeaite. Krennerite Au$_3$AgTe$_8$ (Dye and Smith, 2012) has TeAu$_3$ pyramids similar to those of honeaite, but less regular.

**Discussion**

**Paragenetic position of honeaite**

The earliest preserved mineral assemblage was formed during amphibolite facies metamorphism and is calcic plagioclase, hornblende and ilmenite (omission). Ductile deformation is recorded in the orientated ilmenite crystals whereas the randomly oriented sheaves of hornblende indicates continuing crystallization after ductile deformation ceased.

The nature and source of the hydrothermal fluids responsible for the subsequent alterations and mineralization are unknown but the changes from calcic to sodic plagioclase and hornblende to actinolite to prehnite reflect falling temperatures and a transition from amphibolite to greenschist and perhaps sub-greenschist facies conditions (Fig. 16). The presence of late brittle prehnite-bearing microfractures shows that these mineralogical changes were accompanied by a change from ductile to brittle deformation, also seen at the deposit scale by late brittle faults.

The mineralization is intimately connected with the period during which prehnite was deposited since 1. the clouds of ore minerals are largely restricted to microfractures and vughs in areas of groundmass prehnite and 2. it postdates this prehnite but predates the final essentially barren cross-cutting prehnite veinlets. Thus, it was deposited from fluids related to the final stages of calc-silicate hydrothermal alteration, which may explain why the deposit scale relationship between all phases of calc-silicate alteration and mineralization is not closer (Fig. 2). The proposed late appearance of the mineralization is not incompatible with other studies that have described it occurring in earlier alteration phases at Karonie (Roberts et al. 2004), since in places these later ore fluids may have migrated outside the calc-silicate envelope.
The close spatial association of honeaite with gold and the other tellurides and especially the presence of composite grains indicate that honeaite is coeval with the rest of the mineralization. The presence of honeaite is further evidence for low temperature conditions since thallium-bearing minerals are typically found in low temperature hydrothermal systems such as Carlin-type gold deposits (Dickson et al. 1979; Cline et al. 2005). The combination of the style of mineralization (vugh- and fracture-hosted) and the associated mineral assemblage all suggest greenschist/ sub-greenschist rather than amphibolite facies conditions and emplacement at relatively shallow crustal levels.

While our limited study clearly points towards deposition of the observed mineralization at relatively low temperatures as noted by de Luca (1995), it may have been remobilised from earlier higher temperature deposits in the Karonie shear zone (Roberts et al. 2004; Jones and Hall 2004).

Comparisons with other telluride-bearing deposits in the Eastern Goldfields and classification of the mineralisation.

The Karonie deposit is one of several gold deposits in the Eastern Goldfields, which contain telluride mineralization that is late tectonic and post peak metamorphism and emplaced in brittle structures. Examples include Sunrise Dam (Baker et al. 2010), Mt Charlotte and Golden Mile (Mueller and Muhling 2013) and Bellerophon Nelson (Xue and Campbell 2014).

Rice et al

However, the close association of the gold and telluride mineralization at Karonie with low temperature calc-silicate (prehnite) alteration appears to be unique.

The Karonie deposit together with most other lode gold deposits in the Eastern Goldfields may be described as an orogenic gold deposit (Goldfarb et al. 2005). Despite the high metamorphic grade (amphibolite) of the host rocks the association of the gold and telluride mineralization at Karonie with brittle structures, vughs and prehnite alteration would place it towards the upper rather than lower crustal end member of the Continuum Model for orogenic gold deposits (Groves 1993). The transition from high temperature (amphibolite facies) to low temperature (greenschist facies) mineral assemblages could be explained by rapid late orogenic exhumation accompanied by reactivation of the Karonie shear zone (Groves et al. 1987; Vielreicher et al. 2015).

Acknowledgements

Ted Wilton (Radium Trail Geoscience, Nevada, USA) (words omitted) for providing additional core samples containing honeaite. Jenny Johnston (School of Geosciences, University of Aberdeen) for diagrams.

An anonymous reviewer for alerting us to the existence of some synthetic compounds which have an analogous stoichiometry to honeaite.
CJS acknowledges Natural Environment Research Council grant NE/M010848/1 Tellurium and Selenium Cycling and Supply.

References


Ref omitted


Rice et al


Ref omitted

Ref omitted


**Figure captions**

1. Regional location of the Karonie gold deposit.
2. Geological map of the Karonie Main Zone gold deposit at the 275m level (simplified and taken from Roberts et al., 2004, after Poseidon Exploration Ltd., 1992
3. A. False colour image of polished thin section of ore sample KD 41/78. Orange = epidote + prehnite; blue = plagioclase feldspar, mainly sodic plagioclase; green = hornblende; white specks = ore minerals. Metamorphic banding east-west. NW-SE veinlets are prehnite only. B. Same section normal colour, green = mainly hornblende, white = prehnite + epidote.
4. BSEM image (KD 41 78m)(4/21) Hornblende being replaced by epidote and actinolite (two generations, dark=actinolite, light= ferro-actinolite). Prehnite mantling
actinolite and filling cavities left by dissolution of plagioclase? Gold (white) occurring in cavities and fractures.

5. BSEM image (KD 41 78m)(4/20) Prehnite filling fractures in epidote. Gold hosted by prehnite. Ilmenite partially replaced by titanite.

6. BSEM image (KD 41 78m)(4/23) Late prehnite occurring in a brittle fracture cutting early prehnite and hornblende. The early prehnite contains numerous small white inclusions of a gold telluride (AuTe), whereas the late prehnite vein is barren. Ilmenite is altered to titanite.

7. BSEM image (KD 41 78m)(7/9) Typical ‘cloud’ of mineralization hosted by prehnite, which is replacing plagioclase. The white inclusions are mainly gold and tellurobismuthite (BiTe).

8. BSEM image (KD 41 78m)(12/2) Composite grain of gold, a bismuth telluride and a nickel cobalt telluride hosted by prehnite. PbTe = a lead telluride.

9. BSEM image (KD 41 78m)(1/11A) Composite grain of honeaite, a bismuth telluride and gold hosted by prehnite.

10. BSEM image (KD 41 78m)(4/1) Honeaite in vugh within prehnite.

11. BSEM image (KD 41 78m)(4/8) Honeaite partially filling cavity in prehnite with a bismuth telluride (BiTe). Gold occurs in a fracture partially exploited by honeaite.

12. BSEM image (KD 41 78m)(4/14) Honeaite occurring in cavity in zoned prehnite.

13. Reflected light digital image of honeaite (bluish grey) (same honeaite grain as in fig. 12), gold (yellow) and pyrrhotite (brownish-pink) in a gangue of prehnite.

14. Reflectance spectra for honeaite

15. Crystal structure of honeaite. (a) A single corrugated sheet of corner-linked TeAu$_3$ pyramids comprising six-membered rings. (b) Full structure showing two double-sheets with TeAu$_3$, pyramids shown in grey. Thallium atoms (green) occur in grooves in the corrugated sheets. The unit-cell projection is shown as a dotted red line. (c) Thallium coordination environment in which there are three Ti-Au intra-sheet bonds and a single Ti-Au inter-sheet bond (blue Au label). There are two long Ti–Te intra-sheet distances (dotted lines) that may not be genuine bonds. Numbers are bond distances in Å. (d) Ball-and-spoke representation of (b) showing intra-sheet Au-Au bonds (solid blue lines), Ti-Au bonds (black dashed lines) and Ti–Te distances (dashed red lines).

16. The paragenesis of honeaite

Tables

1. Electron microprobe analyses of Karonie silicates, gold and tellurides.

Footnote: The analyses were carried out using a MICROSCAN MKS in energy dispersive mode (Link Analytical AN10/2SS). The instrumental conditions were: accelerating voltage 15kV, current 2.8 nA, beam diameter c. 5 microns, take off angle 75 degrees and livetime 200 seconds. The standards used are a mixture of natural minerals, metal oxides and pure metals.

2. Reflectance data in air and in oil for honeaite. (COM refers to the Commission on Ore Mineralogy recommended minimum wavelengths)

3. Electron microprobe analyses (n=17) of honeaite.

Footnote: Standards used for honeaite: Au = Au metal; TI = Thallium iodide; Te = Te metal.

Instrument and operating conditions in Table 1.
4. Calculated powder XRD pattern of honeaite. Only reflections with relative intensities ≥ 5% are shown.
Fig. 1. Regional location of the Karonie gold deposit.

Fig. 2. Geological map of the Karonie Main Zone gold deposit at the 275m level (simplified and taken from Roberts et al., 2004, after Poseidon Exploration Ltd., 1992)
Fig. 3. A. False colour image of polished thin section of ore **sample KD 41/78**. Orange = epidote + prehnite; blue = plagioclase feldspar, mainly sodic plagioclase; green = hornblende; white specks = ore minerals. Metamorphic banding east-west. NW-SE veinlets are prehnite only. B. Same section normal colour, green = mainly hornblende, white = prehnite + epidote.

Fig. 4. BSEM image (KD 41 78m) (4/21) Hornblende being replaced by epidote and actinolite (two generations, dark=actinolite, light=ferro-actinolite). Prehnite mantling actinolite and filling cavities left by dissolution of plagioclase? Gold (white) occurring in cavities and fractures.
Fig. 5. BSEM image (KD 41 78m) (4/20) Prehnite filling fractures in epidote. Gold hosted by prehnite. Ilmenite partially replaced by titanite.

Fig. 6. BSEM image (KD 41 78m) (4/23) Late prehnite occurring in a brittle fracture cutting early prehnite and hornblende. The early prehnite contains numerous small white inclusions of a gold telluride (AuTe), whereas the late prehnite vein is barren. Ilmenite is altered to titanite.
Fig. 7. BSEM image (KD 41 78m) (7/9) Typical ‘cloud’ of mineralization hosted by prehnite, which is replacing plagioclase. The white inclusions are mainly gold and tellurobismuthite (BiTe).

Fig. 8. BSEM image (KD 41 78m) (12/2) Composite grain of gold, a bismuth telluride and a nickel cobalt telluride hosted by prehnite. PbTe = a lead telluride.
Fig. 9. BSEM image (KD 41 78m) (1/11A) Composite grain of honeaite, a bismuth telluride and gold hosted by prehnite.

Fig. 10. BSEM image (KD 41 78m) (4/1) Honeaite in vugh within prehnite.
Fig. 11. BSEM image (KD 41 78m) (4/8) Honeaite partially filling cavity in prehnite with a bismuth telluride (BiTe). Gold occurs in a fracture partially exploited by honeaite.

Fig. 12. BSEM image (KD 41 78m) (4/14) Honeaite occurring in cavity in zoned prehnite.
Fig. 13. Reflected light digital image of honeaite (bluish grey) (same honeaite grain as in fig. 12), gold (yellow) and pyrrhotite (brownish-pink) in a gangue of prehnite.

Fig. 14. Reflectance spectra for honeaite
Fig. 15. Crystal structure of honeaite. (a) A single corrugated sheet of corner-linked TeAu₃ pyramids comprising six-membered rings. (b) Full structure showing two double-sheets with TeAu₃ pyramids shown in grey. Thallium atoms (green) occur in grooves in the corrugated sheets. The unit-cell projection is shown as a dotted red line. (c) Thallium coordination environment in which there are three Tl-Au intra-sheet bonds and a single Tl-Au inter-sheet bond (blue Au label). There are two long Tl–Te intra-sheet distances (dotted lines) that may not be genuine bonds. Numbers are bond distances in Å. (d) Ball-and-spoke representation of (b) showing intra-sheet Au-Au bonds (solid blue lines), Tl-Au bonds (black dashed lines) and Tl–Te distances (dashed red lines).
Fig. 16. The paragenesis of honeaite.
<table>
<thead>
<tr>
<th></th>
<th>Hornblende</th>
<th>Ferro-actinolite</th>
<th>Actinolite</th>
<th>Epidote</th>
<th>Prehnite</th>
<th>Plagioclase</th>
</tr>
</thead>
<tbody>
<tr>
<td>SiO₂</td>
<td>43.74</td>
<td>43.84</td>
<td>43.14</td>
<td>51.59</td>
<td>51.95</td>
<td>54.83</td>
</tr>
<tr>
<td>TiO₂</td>
<td>0.51</td>
<td>0.53</td>
<td>0.63</td>
<td>0.00</td>
<td>0.00</td>
<td>0.02</td>
</tr>
<tr>
<td>Al₂O₃</td>
<td>12.48</td>
<td>11.86</td>
<td>12.56</td>
<td>0.88</td>
<td>0.41</td>
<td>0.45</td>
</tr>
<tr>
<td>FeO</td>
<td>20.15</td>
<td>20.18</td>
<td>20.30</td>
<td>27.04</td>
<td>26.41</td>
<td>15.80</td>
</tr>
<tr>
<td>Fe₂O₃</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>6.55</td>
<td>10.23</td>
<td>1.96</td>
</tr>
<tr>
<td>MnO</td>
<td>0.33</td>
<td>0.35</td>
<td>0.33</td>
<td>0.35</td>
<td>0.29</td>
<td>0.08</td>
</tr>
<tr>
<td>MgO</td>
<td>7.70</td>
<td>7.88</td>
<td>7.62</td>
<td>5.86</td>
<td>6.92</td>
<td>13.79</td>
</tr>
<tr>
<td>CaO</td>
<td>11.65</td>
<td>11.43</td>
<td>11.30</td>
<td>12.11</td>
<td>12.39</td>
<td>12.64</td>
</tr>
<tr>
<td>Na₂O</td>
<td>1.76</td>
<td>1.63</td>
<td>1.64</td>
<td>0.20</td>
<td>0.21</td>
<td>0.18</td>
</tr>
<tr>
<td>K₂O</td>
<td>0.28</td>
<td>0.24</td>
<td>0.31</td>
<td>0.03</td>
<td>0.01</td>
<td>0.00</td>
</tr>
<tr>
<td>Total</td>
<td>98.60</td>
<td>97.94</td>
<td>97.82</td>
<td>98.06</td>
<td>98.58</td>
<td>97.80</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th></th>
<th>Si</th>
<th>Ti</th>
<th>Al</th>
<th>Fe²⁺</th>
<th>Fe³⁺</th>
<th>Mn</th>
<th>Mg</th>
<th>Ca</th>
<th>Na</th>
<th>K</th>
<th>Total</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>6.56</td>
<td>0.06</td>
<td>2.21</td>
<td>2.53</td>
<td>-</td>
<td>0.04</td>
<td>1.72</td>
<td>1.87</td>
<td>0.51</td>
<td>0.05</td>
<td>15.56</td>
</tr>
<tr>
<td></td>
<td>6.62</td>
<td>0.06</td>
<td>2.11</td>
<td>2.55</td>
<td>-</td>
<td>0.05</td>
<td>1.77</td>
<td>1.85</td>
<td>0.48</td>
<td>0.06</td>
<td>15.53</td>
</tr>
<tr>
<td></td>
<td>6.53</td>
<td>0.07</td>
<td>2.24</td>
<td>2.57</td>
<td>-</td>
<td>0.04</td>
<td>1.72</td>
<td>1.83</td>
<td>0.48</td>
<td>0.06</td>
<td>15.54</td>
</tr>
<tr>
<td></td>
<td>7.93</td>
<td>0.00</td>
<td>0.16</td>
<td>3.48</td>
<td>-</td>
<td>0.04</td>
<td>1.57</td>
<td>1.99</td>
<td>0.06</td>
<td>0.01</td>
<td>15.01</td>
</tr>
<tr>
<td></td>
<td>7.93</td>
<td>0.00</td>
<td>0.07</td>
<td>3.37</td>
<td>-</td>
<td>0.04</td>
<td>1.57</td>
<td>2.03</td>
<td>0.06</td>
<td>0.01</td>
<td>15.07</td>
</tr>
<tr>
<td></td>
<td>7.98</td>
<td>0.00</td>
<td>0.08</td>
<td>1.92</td>
<td>-</td>
<td>0.04</td>
<td>1.57</td>
<td>1.97</td>
<td>0.05</td>
<td>0.01</td>
<td>15.05</td>
</tr>
<tr>
<td></td>
<td>7.94</td>
<td>0.00</td>
<td>0.11</td>
<td>2.26</td>
<td>-</td>
<td>0.04</td>
<td>1.57</td>
<td>1.98</td>
<td>0.07</td>
<td>0.01</td>
<td>15.01</td>
</tr>
<tr>
<td></td>
<td>3.00</td>
<td>0.00</td>
<td>0.01</td>
<td>-</td>
<td>0.38</td>
<td>0.04</td>
<td>0.01</td>
<td>0.02</td>
<td>0.01</td>
<td>0.00</td>
<td>8.01</td>
</tr>
<tr>
<td></td>
<td>3.03</td>
<td>0.00</td>
<td>0.03</td>
<td>0.61</td>
<td>0.38</td>
<td>0.04</td>
<td>0.01</td>
<td>0.02</td>
<td>0.01</td>
<td>0.00</td>
<td>8.00</td>
</tr>
<tr>
<td></td>
<td>6.09</td>
<td>0.00</td>
<td>0.01</td>
<td>0.21</td>
<td>0.61</td>
<td>0.04</td>
<td>0.01</td>
<td>0.03</td>
<td>0.01</td>
<td>0.00</td>
<td>13.96</td>
</tr>
<tr>
<td></td>
<td>6.01</td>
<td>0.00</td>
<td>0.00</td>
<td>0.03</td>
<td>0.21</td>
<td>0.04</td>
<td>0.01</td>
<td>0.03</td>
<td>0.01</td>
<td>0.00</td>
<td>14.01</td>
</tr>
<tr>
<td></td>
<td>10.71</td>
<td>0.00</td>
<td>0.00</td>
<td>0.01</td>
<td>0.03</td>
<td>0.04</td>
<td>0.01</td>
<td>0.03</td>
<td>0.01</td>
<td>0.00</td>
<td>20.03</td>
</tr>
<tr>
<td></td>
<td>10.75</td>
<td>0.00</td>
<td>0.00</td>
<td>0.04</td>
<td>0.04</td>
<td>0.04</td>
<td>0.01</td>
<td>0.03</td>
<td>0.01</td>
<td>0.00</td>
<td>19.98</td>
</tr>
<tr>
<td></td>
<td>11.66</td>
<td>0.00</td>
<td>0.00</td>
<td>0.01</td>
<td>0.04</td>
<td>0.04</td>
<td>0.01</td>
<td>0.03</td>
<td>0.01</td>
<td>0.00</td>
<td>19.95</td>
</tr>
<tr>
<td></td>
<td>11.57</td>
<td>0.00</td>
<td>0.00</td>
<td>0.01</td>
<td>0.04</td>
<td>0.04</td>
<td>0.01</td>
<td>0.03</td>
<td>0.01</td>
<td>0.00</td>
<td>19.92</td>
</tr>
</tbody>
</table>

<p>| An %     | 32.58      | 32.70   | 8.79    | 11.02  |</p>
<table>
<thead>
<tr>
<th></th>
<th>Gold</th>
<th>Petzite</th>
<th>Calaverite</th>
<th>Hessite</th>
<th>Tellurobismuthite</th>
<th>Melonite</th>
<th>Mattagamite</th>
<th>Frohbergite</th>
</tr>
</thead>
<tbody>
<tr>
<td>Au</td>
<td>94.30</td>
<td>94.22</td>
<td>25.74</td>
<td>26.17</td>
<td>0.00</td>
<td>Fe</td>
<td>0.00</td>
<td>0.09</td>
</tr>
<tr>
<td>Ag</td>
<td>5.66</td>
<td>5.71</td>
<td>40.45</td>
<td>40.08</td>
<td>0.59</td>
<td>0.60</td>
<td>62.47</td>
<td>Co</td>
</tr>
<tr>
<td>Bi</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>52.38</td>
<td>52.59</td>
</tr>
<tr>
<td>Te</td>
<td>-</td>
<td>-</td>
<td>33.58</td>
<td>33.42</td>
<td>57.09</td>
<td>56.35</td>
<td>38.34</td>
<td>47.36</td>
</tr>
<tr>
<td>Total</td>
<td>99.96</td>
<td>99.93</td>
<td>99.77</td>
<td>99.67</td>
<td>100.59</td>
<td>99.60</td>
<td>100.81</td>
<td>99.74</td>
</tr>
</tbody>
</table>

\[
\text{Te} = 2.00 
\]

<table>
<thead>
<tr>
<th></th>
<th>Au</th>
<th>Ag</th>
<th>Bi</th>
<th>Te</th>
<th></th>
<th>Fe</th>
<th>Co</th>
<th>Ni</th>
<th>Te</th>
</tr>
</thead>
<tbody>
<tr>
<td>Au</td>
<td>0.99</td>
<td>1.01</td>
<td>2.85</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
</tr>
<tr>
<td>Ag</td>
<td>2.85</td>
<td>2.84</td>
<td>0.02</td>
<td>0.03</td>
<td>1.93</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
</tr>
<tr>
<td>Bi</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>2.03</td>
<td>2.04</td>
<td>-</td>
<td>-</td>
</tr>
<tr>
<td>Te</td>
<td>2.00</td>
<td>2.00</td>
<td>2.00</td>
<td>2.00</td>
<td>2.00</td>
<td>2.00</td>
<td>2.00</td>
<td>2.00</td>
<td>2.00</td>
</tr>
</tbody>
</table>

Table 1. Electron microprobe analyses of Karonie silicates, gold and tellurides

The analyses were carried out using a MICROSCAN MK5 in energy dispersive mode (Link Analytical AN10/25S). The instrumental conditions were: accelerating voltage 15kV, current 2.8 nA, beam diameter c. 5 microns, take off angle 75 degrees and livetime 200 seconds. The standards used are a mixture of natural minerals, metal oxides and pure metals.
Table 2. Reflectance data in air and in oil for honeaite. (COM refers to the Commission on Ore Mineralogy recommended minimum wavelengths)

<table>
<thead>
<tr>
<th>$R_1$</th>
<th>$R_2$</th>
<th>$^\text{im}R_1$</th>
<th>$^\text{im}R_2$</th>
<th>$\lambda$/nm</th>
<th>$R_1$</th>
<th>$R_2$</th>
<th>$^\text{im}R_1$</th>
<th>$^\text{im}R_2$</th>
<th>$\lambda$/nm</th>
</tr>
</thead>
<tbody>
<tr>
<td>52.4</td>
<td>50.0</td>
<td>38.7</td>
<td>36.5</td>
<td>400</td>
<td>50.9</td>
<td>51.4</td>
<td>36.6</td>
<td>37.7</td>
<td>560</td>
</tr>
<tr>
<td>52.6</td>
<td>50.3</td>
<td>38.7</td>
<td>36.9</td>
<td>420</td>
<td>50.8</td>
<td>51.7</td>
<td>36.4</td>
<td>38.0</td>
<td>580</td>
</tr>
<tr>
<td>52.5</td>
<td>50.7</td>
<td>38.3</td>
<td>37.2</td>
<td>440</td>
<td>50.7</td>
<td>51.8</td>
<td>36.3</td>
<td>38.0</td>
<td>589 (COM)</td>
</tr>
<tr>
<td>51.9</td>
<td>50.9</td>
<td>37.8</td>
<td>37.4</td>
<td>460</td>
<td>50.6</td>
<td>51.9</td>
<td>36.1</td>
<td>38.1</td>
<td>600</td>
</tr>
<tr>
<td>51.7</td>
<td>50.9</td>
<td>37.6</td>
<td>37.4</td>
<td>470 (COM)</td>
<td>50.5</td>
<td>51.8</td>
<td>36.0</td>
<td>38.2</td>
<td>620</td>
</tr>
<tr>
<td>51.5</td>
<td>50.9</td>
<td>37.4</td>
<td>37.4</td>
<td>480</td>
<td>50.3</td>
<td>51.4</td>
<td>35.7</td>
<td>37.7</td>
<td>640</td>
</tr>
<tr>
<td>51.3</td>
<td>51.0</td>
<td>37.3</td>
<td>37.4</td>
<td>500</td>
<td>50.2</td>
<td>51.2</td>
<td>35.5</td>
<td>37.5</td>
<td>650 (COM)</td>
</tr>
<tr>
<td>51.3</td>
<td>51.3</td>
<td>36.9</td>
<td>37.4</td>
<td>520</td>
<td>50.1</td>
<td>50.9</td>
<td>35.4</td>
<td>37.2</td>
<td>660</td>
</tr>
<tr>
<td>51.2</td>
<td>51.3</td>
<td>36.8</td>
<td>37.6</td>
<td>540</td>
<td>49.8</td>
<td>50.3</td>
<td>35.0</td>
<td>36.6</td>
<td>680</td>
</tr>
<tr>
<td>51.1</td>
<td>51.3</td>
<td>36.7</td>
<td>37.6</td>
<td>546 (COM)</td>
<td>49.5</td>
<td>50.0</td>
<td>34.9</td>
<td>36.0</td>
<td>700</td>
</tr>
</tbody>
</table>
Table 3. Electron microprobe analyses of honeaite

<table>
<thead>
<tr>
<th>Honeaite</th>
<th>Au</th>
<th>Tl</th>
<th>Te</th>
<th>Total</th>
</tr>
</thead>
<tbody>
<tr>
<td>Crystal.1</td>
<td>56.24</td>
<td>19.65</td>
<td>24.40</td>
<td>100.29</td>
</tr>
<tr>
<td>Crystal.2</td>
<td>55.82</td>
<td>19.83</td>
<td>24.24</td>
<td>99.88</td>
</tr>
<tr>
<td>Crystal.3</td>
<td>56.19</td>
<td>19.45</td>
<td>24.19</td>
<td>99.82</td>
</tr>
<tr>
<td>Crystal.4</td>
<td>56.34</td>
<td>19.66</td>
<td>24.25</td>
<td>100.25</td>
</tr>
<tr>
<td>Crystal.5</td>
<td>55.91</td>
<td>19.96</td>
<td>24.30</td>
<td>100.16</td>
</tr>
<tr>
<td>Crystal.6</td>
<td>56.73</td>
<td>19.49</td>
<td>24.17</td>
<td>100.39</td>
</tr>
<tr>
<td>Crystal.7</td>
<td>56.77</td>
<td>19.83</td>
<td>24.38</td>
<td>100.98</td>
</tr>
<tr>
<td>Crystal.8</td>
<td>56.43</td>
<td>19.11</td>
<td>24.32</td>
<td>99.86</td>
</tr>
<tr>
<td>Crystal.9</td>
<td>56.88</td>
<td>19.86</td>
<td>24.27</td>
<td>101.02</td>
</tr>
<tr>
<td>Crystal.10</td>
<td>56.53</td>
<td>20.10</td>
<td>24.22</td>
<td>100.86</td>
</tr>
<tr>
<td>Crystal.11</td>
<td>56.47</td>
<td>19.70</td>
<td>24.30</td>
<td>100.47</td>
</tr>
<tr>
<td>Crystal.12</td>
<td>56.79</td>
<td>19.23</td>
<td>24.09</td>
<td>100.10</td>
</tr>
<tr>
<td>Crystal.13</td>
<td>55.83</td>
<td>19.99</td>
<td>24.59</td>
<td>100.41</td>
</tr>
<tr>
<td>Crystal.14</td>
<td>56.61</td>
<td>19.78</td>
<td>24.44</td>
<td>100.83</td>
</tr>
<tr>
<td>Crystal.15</td>
<td>55.75</td>
<td>19.82</td>
<td>24.33</td>
<td>99.90</td>
</tr>
<tr>
<td>Crystal.16</td>
<td>56.33</td>
<td>19.79</td>
<td>24.34</td>
<td>100.46</td>
</tr>
<tr>
<td>Crystal.17</td>
<td>55.98</td>
<td>19.31</td>
<td>24.31</td>
<td>99.60</td>
</tr>
</tbody>
</table>

Constituent | Wt.% (n=17) | Range   | Stand. Dev. (1σ) |
------------|-------------|---------|------------------|
Au          | 56.33       | 55.75-56.88 | 0.37             |
Tl          | 19.68       | 19.11-20.10 | 0.27             |
Te          | 24.30       | 24.09-24.59 | 0.11             |
Total       | 100.31      |         |                  |

Standards used for honeaite: Au = Au metal; Tl = Thallium iodide; Te = Te metal.
Instrument and operating conditions in Table 1.
Table 4 Calculated powder XRD pattern of honeaite. Only reflections with relative intensities ≥ 5% are shown.

<table>
<thead>
<tr>
<th>h k l</th>
<th>d_{hkl} (Å)</th>
<th>I/I_{max} (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>1 1 1</td>
<td>4.915</td>
<td>13</td>
</tr>
<tr>
<td>0 2 0</td>
<td>4.438</td>
<td>6</td>
</tr>
<tr>
<td>1 2 1</td>
<td>3.547</td>
<td>6</td>
</tr>
<tr>
<td>3 0 0</td>
<td>2.989</td>
<td>31</td>
</tr>
<tr>
<td>0 2 2</td>
<td>2.938</td>
<td>100</td>
</tr>
<tr>
<td>3 1 0</td>
<td>2.833</td>
<td>23</td>
</tr>
<tr>
<td>3 1 2</td>
<td>2.296</td>
<td>14</td>
</tr>
<tr>
<td>0 4 0</td>
<td>2.219</td>
<td>15</td>
</tr>
<tr>
<td>3 2 2, 4 1 1</td>
<td>2.095</td>
<td>47 (39, 8)</td>
</tr>
<tr>
<td>0 0 4</td>
<td>1.960</td>
<td>16</td>
</tr>
<tr>
<td>3 3 2</td>
<td>1.853</td>
<td>17</td>
</tr>
<tr>
<td>3 4 0</td>
<td>1.782</td>
<td>8</td>
</tr>
<tr>
<td>3 0 4</td>
<td>1.639</td>
<td>9</td>
</tr>
<tr>
<td>3 1 4</td>
<td>1.612</td>
<td>7</td>
</tr>
<tr>
<td>0 4 4</td>
<td>1.469</td>
<td>7</td>
</tr>
<tr>
<td>3 4 4</td>
<td>1.319</td>
<td>5</td>
</tr>
<tr>
<td>6 3 2</td>
<td>1.263</td>
<td>7</td>
</tr>
<tr>
<td>0 2 6</td>
<td>1.254</td>
<td>5</td>
</tr>
</tbody>
</table>